

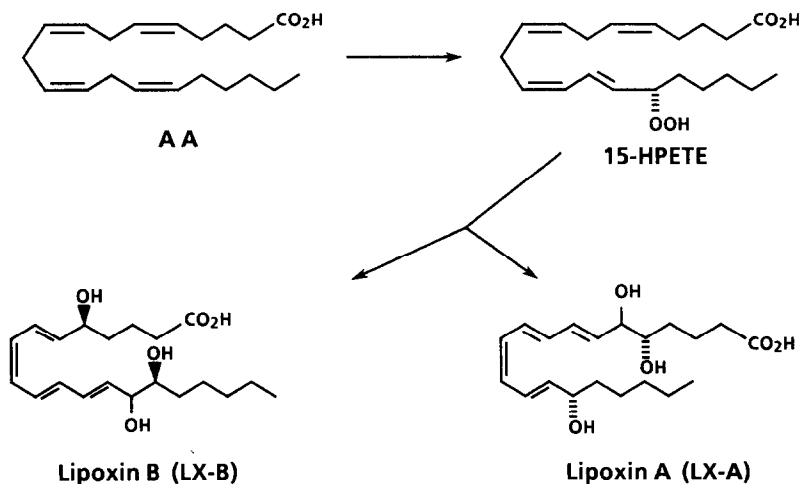
SYNTHESIS OF LIPOXIN B

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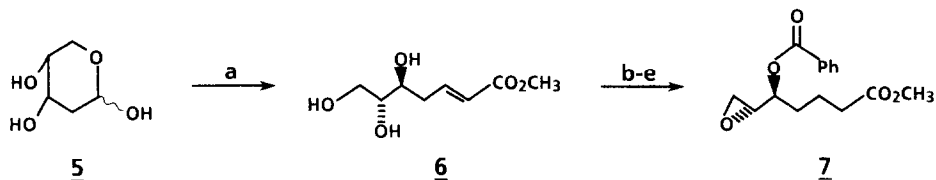
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ABSTRACT: *The synthesis of four isomers of 5,14,15-triHETE is reported. The key step is the coupling of acetylene 11 (or 19) with epoxide 7 to afford alcohol 12 (or 20). Natural lipoxin B is found to correspond to a mixture of 14R-8-cis-1, 14R-8-trans-3 and 14S-8-trans-4.*

While studying the relationship between the 5- and 15-lipoxygenase pathways in 1983, Serhan, Hamberg and Samuelsson isolated a novel series of oxygenated arachidonic acid derivatives produced when human leukocytes were treated with exogenous 15-hydroperoxy-5,8,11,13-eicosatetraenoic acid (15-HPETE) in the presence of the calcium ionophore A-231871. Initially, two generic isomeric structures were identified from these studies, namely 5,6,15-trihydroxy-7,9,11,13-eicosatetraenoic acid (5,6,15-triHETE) and 5,14,15-trihydroxy-6,8,10,12-eicosatetraenoic acid (5,14,15-triHETE). Because of their potential formation via multiple lipoxygenation reactions, these metabolites were given the trivial names lipoxin A (LX-A) and lipoxin B (LX-B), respectively. Encouraged by reports of interesting biological activities for these compounds^{1,2}, much attention has focused on the elucidation of their absolute stereochemistry and olefin geometry^{3,4}. With regards to lipoxin B, two different structural assignments have been made^{4a,d}. In this paper, we disclose our findings in this area and report the synthesis of four isomers of 5,14,15-triHETE. In contrast with both of the previous claims we find that natural lipoxin B corresponds to a mixture of three of these synthetic isomers.



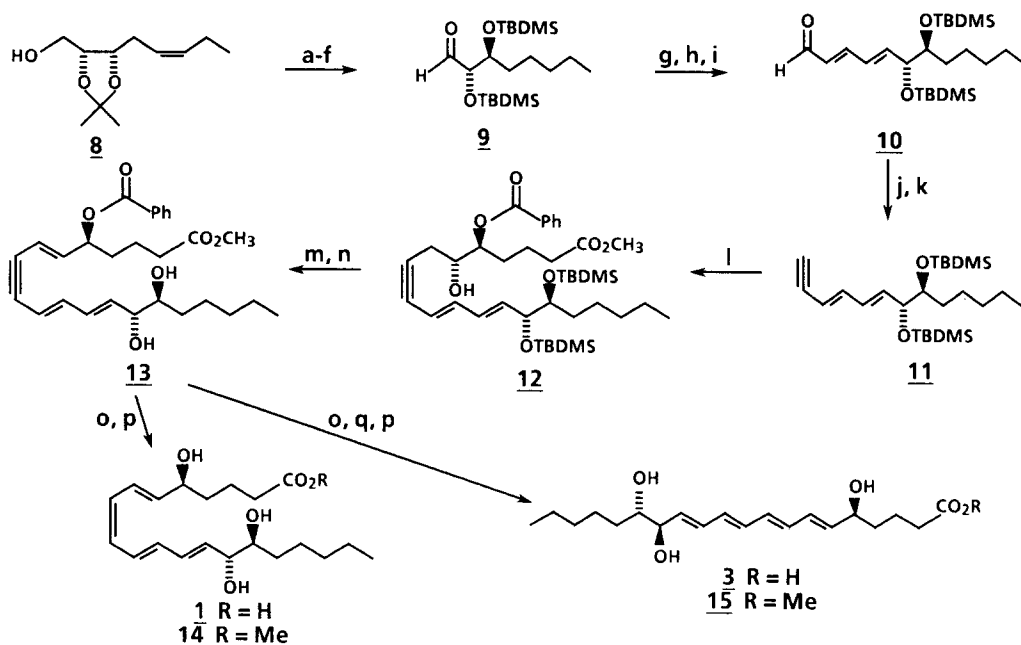
In choosing specific isomers of 5,14,15-triHETE for total synthesis, we considered a variety of potential biosynthetic pathways for the formation of lipoxin B, all of which assume an initial 5-lipoxygenation of 15-HPETE to produce 5S,15S-diHPETE as a common intermediate⁵. On the basis of this analysis four isomers were targeted for total synthesis, namely, 5S,14R,15S-6,10,12-*trans*-8-*cis*-triHETE (**1**), 5S,14S,15S-6,10,12-*trans*-8-*cis*-triHETE (**2**) and the corresponding all *trans* derivatives **3** and **4**, respectively. Our strategy was to assemble these molecules via the coupling of an appropriately functionalized C₁-C₇ epoxide with a suitably protected conjugated acetylene (C₈-C₂₀) in which the hydroxyl stereocenters at C-5,14 and 15 were derived from optically pure carbohydrates. Epoxide **7** was synthesized in 5 steps starting from 2-deoxy-D-ribose (**5**) (scheme 1). This compound, previously described as an intermediate in total syntheses of leukotrienes A₄ and B₄⁶, was the C₁-C₇ fragment for all of the lipoxin B isomers.



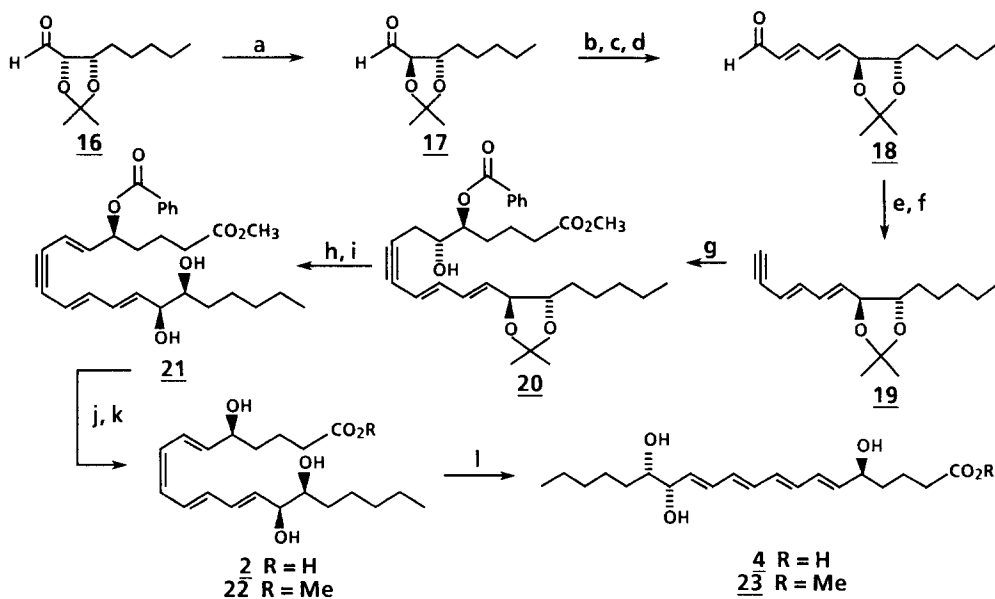
SCHEME 117: a. $\text{Ph}_3\text{P}=\text{CHCO}_2\text{Me}$ b. $\text{C}_9\text{H}_{11}\text{SO}_2\text{Cl}$, pyr. c. H_2 , Pd/C d. K_2CO_3 , MeOH e. PhCOCl , pyr.

Compound **5** was also the starting material for the synthesis of the C₈-C₂₀ acetylenes **11** and **19**. Wittig reaction of the acetonide of 2-deoxy-D-ribose with the ylid derived from $n\text{PrPh}_3\text{P}^+\text{Br}^-$ afforded alcohol **8** (82%) which was converted to aldehyde **9** in 69% overall yield (scheme 2)⁷. Treatment of **9** with the lithium anion of *cis*-1-methoxy-1-buten-3-yne (THF, -78°C) afforded the corresponding propargylic alcohol (99%). Reduction with Red-Al (PhCH_3 , -78 to 0°C)^{9,10} gave the diene (*cis*, *trans*) which upon treatment with MsCl (Et_3N , THF, -40°C) followed by aqueous hydrolysis afforded dienealdehyde **10** in 30% yield⁸. Wittig reaction of **10** with the ylid prepared from $\text{ClCH}_2\text{Ph}_3\text{P}^-\text{Cl}^-$ (THF, 0°C) gave a mixture of chloroolefins (96%) which when exposed to LDA (THF, 0°C) produced acetylene **11** (87%). Following the conditions described by Yamaguchi¹¹, successive treatment of acetylene **11** with $n\text{BuLi}$ and $\text{BF}_3\cdot\text{OEt}_2$ followed by the addition of epoxide **7** (THF, -78°C) afforded a 60% yield of alcohol **12** along with 23% of unreacted **11**. Reaction of **12** with MsCl (Et_3N , THF, -10 to 23°C) followed by treatment of the corresponding mesylate *in situ* with DBU gave an 8:1 mixture of 6-*trans*/6-*cis* olefins in which the major isomer was isolated in 57% yield (>98% purity). Deprotection with $n\text{Bu}_4\text{NF}$ (3 eq. THF, 25°C) afforded diol **13** in 73% yield. Semihydrogenation of **13** was carried out using 5% Pd/ CaCO_3 (Lindlar) in 5% pyridine/ EtOAc to give a 50% yield of the 8-*cis* tetraene after purification by RP-HPLC. Removal of the benzoate with K_2CO_3 in methanol afforded methyl ester **14**. Alternatively, hydrolysis of the benzoate methyl ester with LiOH followed by acidification to pH 3.5 with aqueous KHSO_4 gave acid **1** (89%). Isomerization of the 8-*cis* olefin with iodine (CH_2Cl_2) afforded the corresponding all *trans* benzoate methyl ester. Hydrolysis as above gave the all *trans* methyl ester **15** and acid **3**, respectively.

In order to use this same methodology to synthesize the corresponding 14S isomers, acetylene **19** was required. Hydrogenation (5% Pd/C, EtOAc) of alcohol **8** followed by Swern¹² oxidation afforded aldehyde **16** (80%). Following a similar example reported by Sharpless¹³, treatment of **16** with K_2CO_3 in methanol resulted in a clean epimerization of the α -alkoxy center giving aldehyde **17**. Conversion of **17** to dienealdehyde **18** (73%) and on to acetylene **19** (53%) was accomplished by the same sequence as described for the 14R case (scheme 3). Sequential treatment of **19** with $n\text{-BuLi}$ and $\text{BF}_3\cdot\text{OEt}_2$ as before followed by the addition of epoxide **7** gave a 63% yield of the coupled product **20** along with 22% un-



SCHEME 217: a. PhCOCl , pyr. b. H_2 , Pd/C c. aq. HCl d. TBDMsCl, imd., DMF e. Dibal, CH_2Cl_2 , -78°C f. $\text{CrO}_3 \cdot 2$ pyr. g. $\text{LiC}\equiv\text{CCH}=\text{CHOMe}$, THF, -78°C h. Red-Al i. MsCl, Et_3N , H_2O j. $\text{ClCH}_2\text{PPh}_3\text{Cl}$, nBuLi k. LDA, THF, 0°C l. nBuLi, $\text{BF}_3 \cdot \text{OEt}_2$; \mathcal{Z} , -78°C m. MsCl, Et_3N ; DBU n. nBu₄NF o. H_2 , Lindlar, 5% pyr/EtOAc p. K_2CO_3 , MeOH or LiOH; KHSO_4 q. I_2 , CH_2Cl_2



SCHEME 317: a. K_2CO_3 , MeOH b. $\text{LiC}\equiv\text{CCH}=\text{CHOMe}$ c. Red-Al d. MsCl, Et_3N ; H_2O e. $\text{ClCH}_2\text{PPh}_3\text{Cl}$, nBuLi f. LDA, THF, 0°C g. nBuLi, $\text{BF}_3 \cdot \text{OEt}_2$; \mathcal{Z} , -78°C h. MsCl, Et_3N ; DBU i. 80% HOAc, H_2O j. H_2 , Lindlar, 5% pyr/EtOAc k. K_2CO_3 , MeOH or LiOH; KHSO_4 l. I_2 , CH_2Cl_2

reacted **19**. Elimination of the C-6 hydroxyl group (MsCl, THF; DBU) gave the enyne (85%) as a 9:1 mixture of 6-*trans*/6-*cis* isomers. Removal of the acetonide with 80% aqueous acetic acid (55°C, 84%) afforded diol **21**. Semihydrogenation as above produced a 30% yield of the desired 8-*cis* tetraene after purification by RP-HPLC. Hydrolysis with LiOH followed by acidification to pH 3.3 (aq. KHSO₄) gave acid **2** in 87% yield. Alternatively, the benzoate was removed with K₂CO₃ in methanol to afford methyl ester **22**. Isomerization of **22** with iodine (CH₂Cl₂) produced the corresponding all *trans* methyl ester **23**.

Comparisons of the methyl esters of the synthetic lipoxin B isomers were made by RP-HPLC with a sample of natural lipoxin B methyl ester prepared by Dr. Charles Serhan of the Karolinska Institutet¹⁴. In 65/35 methanol/water (Ultrasphere-ODS, 250x10mm, 3.0mL/min, 300nm) natural lipoxin B appears as a mixture of three components at 34.3, 36.1 and 37.0 min in a relative ratio of 3:5:2, respectively. In coinjection experiments with the synthetic isomers, 5S,14S,15S-6,8,10,12-*trans*-triHETE methyl ester (**23**) was found to coelute with the natural material at 34.3 min. Similarly, 5S,14R,15S-6,8,10,12-*trans*-triHETE methyl ester (**15**) coeluted with the natural material at 36.1 min and 5S,14R,15S-6,10,12-*trans*-8-*cis*-triHETE methyl ester (**14**) matched the third natural component at 37.0 min¹⁵. These results strongly suggest the structures for the natural lipoxin B isomers as indicated¹⁶. The discrepancy between our assignments and those previously reported clearly must reside in the source of the natural material. There is recent evidence to suggest that the 14R-8-*cis* isomer **1** undergoes at least partial isomerization during the isolation process⁵. This may explain the failure of at least one group^{4a,b} from detecting significant amounts of **1** in their sample of natural lipoxin B.

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- We thank Drs: C.N. Serhan and B. Samuelsson for providing a sample of natural lipoxin B.
- By comparison, 5S,14S,15S-6,10,12-*trans*-8-*cis*-triHETE methyl ester (**22**) had a retention time of 61.0 min. under these RP-HPLC conditions.
- The possibility that other 5,14,15-trihydroxytetraenes may have similar retention times cannot be ruled out.
- Satisfactory spectroscopic data were obtained for each reaction product.